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1-Benzoyl-*c*-3,*t*-3-dimethyl-*r*-2,*c*-6-diphenylpiperidin-4-oneS. Aravindhan,^a S. Ponnuswamy,^b J. Umamaheswari,^b
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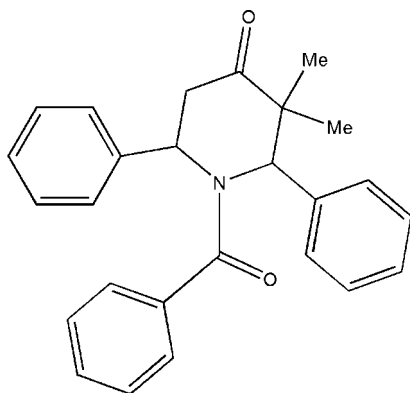
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Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.050; wR factor = 0.152; data-to-parameter ratio = 23.4.

In the title compound, $\text{C}_{26}\text{H}_{25}\text{NO}_2$, the piperidine ring adopts a distorted boat conformation. The three phenyl rings form dihedral angles of 67.58 (8), 59.82 (8) and 86.41 (8) $^\circ$ with the best plane through the piperidine ring. The crystal packing is governed by intermolecular $\text{C}-\text{H}\cdots\text{O}$ interactions.

Related literature

For the biological activity of piperidine derivatives, see: Dimmock *et al.* (2001); Perumal *et al.* (2001). For hydrogen-bond motifs, see: Bernstein *et al.* (1995). For puckering and asymmetry parameters, see: Cremer & Pople (1975); Nardelli (1983).



Experimental

Crystal data

 $\text{C}_{26}\text{H}_{25}\text{NO}_2$ $M_r = 383.47$

Monoclinic, $P2_1/c$
 $a = 10.8540$ (9) Å
 $b = 17.8050$ (17) Å
 $c = 10.8853$ (10) Å
 $\beta = 94.987$ (3) $^\circ$
 $V = 2095.7$ (3) Å³

$Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.08$ mm⁻¹
 $T = 293$ K
 $0.30 \times 0.25 \times 0.20$ mm

Data collection

Bruker Kappa APEXII area-detector diffractometer
 Absorption correction: multi-scan (SADABS; Sheldrick, 2001)
 $T_{\min} = 0.977$, $T_{\max} = 0.985$

27356 measured reflections
 6189 independent reflections
 3897 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.038$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.050$
 $wR(F^2) = 0.152$
 $S = 0.98$
 6189 reflections

265 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.24$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.17$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, $^\circ$).

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|--|-------|-------------|-------------|---------------|
| $\text{C6}-\text{H6}\cdots\text{O1}$ | 0.98 | 2.29 | 2.7346 (17) | 106 |
| $\text{C2}-\text{H2}\cdots\text{O1}^i$ | 0.98 | 2.56 | 3.3784 (17) | 141 |
| $\text{C20}-\text{H20A}\cdots\text{O1}^i$ | 0.96 | 2.47 | 3.1885 (19) | 132 |
| $\text{C20}-\text{H20B}\cdots\text{O2}^{ii}$ | 0.96 | 2.52 | 3.470 (2) | 170 |

Symmetry codes: (i) $x, -y + \frac{1}{2}, z + \frac{1}{2}$; (ii) $x, -y + \frac{1}{2}, z - \frac{1}{2}$.

Data collection: APEX2 (Bruker, 2004); cell refinement: SAINT (Bruker, 2004); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: SHELXL97 and PLATON (Spek, 2009).

SA thanks Dr Babu Varghese, SAIF, IIT-Madras, India, for his help with the data collection.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT2990).

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supplementary materials

Acta Cryst. (2009). E65, o1975 [doi:10.1107/S1600536809028037]

1-Benzoyl-*c*-3,*t*-3-dimethyl-*r*-2,*c*-6-diphenylpiperidin-4-one

S. Aravindhan, S. Ponnuswamy, J. Umamaheswari, P. Ramesh and M. N. Ponnuswamy

Comment

Piperidones are the important group of heterocyclic compounds in the field of medicinal chemistry due to their biological activities, including cytotoxic and anticancer properties (Dimmock *et al.*, 2001). They were also reported to possess analgesic, anti-inflammatory, central nervous system (CNS), local anaesthetic, anticancer and antimicrobial activities (Perumal *et al.*, 2001). In view of these importance and to ascertain the molecular conformation, crystallographic study of the title compound has been carried out.

The *ORTEP* diagram of the title compound is shown in Fig.1. The piperidine ring adopts distorted boat conformation. The puckering parameters (Cremer & Pople, 1975) and the asymmetry parameters (Nardelli, 1983) for this ring are $q_2 = 0.636$ (2) Å, $q_3 = 0.104$ (2) Å, $\pi = 282.8$ (1)° and $\Delta s(C3) = \Delta s(C6) = 18.6$ (1)°. The sum of the angles at N1 (359.7°) is in accordance with sp^2 hybridization. The three phenyl rings are twisted away from the best plane of the piperidine ring by 67.58 (8)°, 59.82 (8)° and 86.41 (8)° respectively.

The crystal packing is controlled by C—H \cdots O types of intra and intermolecular interactions in addition to van der Waals forces. Atom C2 at (*x*, *y*, *z*) donates a proton to O1 *x*, -*y* + 1/2, *z* + 1/2, which forms a C(5) (Bernstein, *et al.*, 1995) zigzag chain running along *c* axis. The combination of C20—H20A \cdots O1 and C20—H20B \cdots O2 intermolecular interactions forms a dimer chain running along *c* axis shown in Fig. 2.

Experimental

A mixture of *c*-3,*t*-3-dimethyl-*r*-2,*c*-6-diphenylpiperidin-4-one (1.4 g, 5 mmol), benzoyl chloride (1.2 ml, 10 mmol) and triethylamine (2 ml, 14.4 mmol) in anhydrous benzene (20 ml) was stirred at room temperature for 7 h. The precipitated ammonium salt was washed with water (4x10ml). The resulting pasty mass was purified and crystallized from benzene and pet-ether (60–80°C) in the ratio of 95: 5.

Refinement

All H atoms were positioned geometrically (C—H=0.93–0.98 Å) and allowed to ride on their parent atoms, with 1.5 $U_{eq}(C)$ for methyl H and 1.2 $U_{eq}(C)$ for other H atoms.

Figures

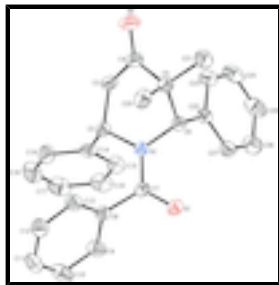


Fig. 1. Perspective view of the molecule showing displacement ellipsoids at 50% probability level. The H atoms are omitted for clarity.

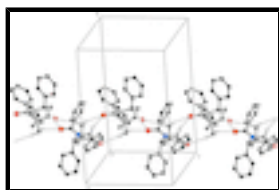


Fig. 2. The crystal packing viewed down *a* axis. H atoms not involved in hydrogen bonding have been omitted for clarity.

1-Benzoyl-*c*-3,*t*-3-dimethyl-*r*-2,*c*-6- diphenylpiperidin-4-one

Crystal data

$C_{26}H_{25}NO_2$

$M_r = 383.47$

Monoclinic, $P2_1/c$

Hall symbol: -P 2ybc

$a = 10.8540$ (9) Å

$b = 17.8050$ (17) Å

$c = 10.8853$ (10) Å

$\beta = 94.987$ (3)°

$V = 2095.7$ (3) Å³

$Z = 4$

$F_{000} = 816$

$D_x = 1.215$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 5746 reflections

$\theta = 1.9$ – 30.4 °

$\mu = 0.08$ mm⁻¹

$T = 293$ K

Block, colorless

$0.30 \times 0.25 \times 0.20$ mm

Data collection

Bruker Kappa APEXII area-detector diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 293$ K

ω and φ scans

Absorption correction: Multi-scan (SADABS; Sheldrick, 2001)

$T_{\min} = 0.977$, $T_{\max} = 0.985$

27356 measured reflections

6189 independent reflections

3897 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.038$

$\theta_{\text{max}} = 30.4$ °

$\theta_{\text{min}} = 1.9$ °

$h = -15 \rightarrow 13$

$k = -25 \rightarrow 25$

$l = -15 \rightarrow 15$

Refinement

| | |
|--|---|
| Refinement on F^2 | Hydrogen site location: inferred from neighbouring sites |
| Least-squares matrix: full | H-atom parameters constrained |
| $R[F^2 > 2\sigma(F^2)] = 0.050$ | $w = 1/[\sigma^2(F_o^2) + (0.0728P)^2 + 0.3353P]$ |
| $wR(F^2) = 0.152$ | where $P = (F_o^2 + 2F_c^2)/3$ |
| $S = 0.98$ | $(\Delta/\sigma)_{\max} = 0.011$ |
| 6189 reflections | $\Delta\rho_{\max} = 0.24 \text{ e } \text{\AA}^{-3}$ |
| 265 parameters | $\Delta\rho_{\min} = -0.17 \text{ e } \text{\AA}^{-3}$ |
| Primary atom site location: structure-invariant direct methods | Extinction correction: SHELXL97 (Sheldrick, 2008), $F_c^* = kF_c[1 + 0.001x F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$ |
| Secondary atom site location: difference Fourier map | Extinction coefficient: 0.0078 (16) |

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | x | y | z | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|-----|---------------|--------------|--------------|----------------------------------|
| O1 | 0.14493 (11) | 0.19520 (7) | 0.50443 (9) | 0.0560 (3) |
| O2 | 0.41301 (12) | 0.16165 (9) | 1.02065 (10) | 0.0722 (4) |
| N1 | 0.17515 (10) | 0.16380 (6) | 0.70591 (9) | 0.0338 (2) |
| C2 | 0.12064 (12) | 0.15019 (8) | 0.82403 (11) | 0.0361 (3) |
| H2 | 0.0881 | 0.1980 | 0.8520 | 0.043* |
| C3 | 0.21997 (14) | 0.12271 (9) | 0.92293 (12) | 0.0466 (4) |
| H3A | 0.1886 | 0.1294 | 1.0030 | 0.056* |
| H3B | 0.2313 | 0.0692 | 0.9113 | 0.056* |
| C4 | 0.34508 (15) | 0.15986 (10) | 0.92630 (13) | 0.0493 (4) |
| C5 | 0.38114 (14) | 0.19299 (9) | 0.80660 (13) | 0.0446 (3) |
| C6 | 0.30890 (12) | 0.15316 (7) | 0.69712 (11) | 0.0362 (3) |
| H6 | 0.3282 | 0.1809 | 0.6235 | 0.043* |
| C7 | 0.10627 (13) | 0.19291 (8) | 0.60735 (11) | 0.0373 (3) |
| C8 | -0.01836 (13) | 0.22524 (8) | 0.62390 (11) | 0.0376 (3) |
| C9 | -0.12156 (16) | 0.19732 (9) | 0.55574 (16) | 0.0543 (4) |
| H9 | -0.1149 | 0.1550 | 0.5067 | 0.065* |

supplementary materials

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|------|---------------|---------------|--------------|------------|
| C10 | -0.23461 (17) | 0.23218 (11) | 0.5604 (2) | 0.0681 (5) |
| H10 | -0.3040 | 0.2125 | 0.5155 | 0.082* |
| C11 | -0.24602 (16) | 0.29517 (10) | 0.62980 (18) | 0.0600 (4) |
| H11 | -0.3223 | 0.3187 | 0.6312 | 0.072* |
| C12 | -0.14437 (17) | 0.32334 (10) | 0.69727 (17) | 0.0588 (4) |
| H12 | -0.1516 | 0.3664 | 0.7445 | 0.071* |
| C13 | -0.03124 (15) | 0.28842 (9) | 0.69574 (15) | 0.0507 (4) |
| H13 | 0.0370 | 0.3075 | 0.7433 | 0.061* |
| C14 | 0.01468 (13) | 0.09444 (8) | 0.80976 (12) | 0.0385 (3) |
| C15 | 0.01631 (15) | 0.03311 (8) | 0.73135 (15) | 0.0485 (4) |
| H15 | 0.0832 | 0.0259 | 0.6848 | 0.058* |
| C16 | -0.08083 (17) | -0.01740 (10) | 0.72184 (18) | 0.0608 (5) |
| H16 | -0.0791 | -0.0582 | 0.6688 | 0.073* |
| C17 | -0.18017 (18) | -0.00735 (12) | 0.7908 (2) | 0.0699 (5) |
| H17 | -0.2455 | -0.0413 | 0.7844 | 0.084* |
| C18 | -0.18221 (18) | 0.05259 (13) | 0.8683 (2) | 0.0724 (6) |
| H18 | -0.2490 | 0.0592 | 0.9152 | 0.087* |
| C19 | -0.08585 (15) | 0.10376 (10) | 0.87796 (15) | 0.0561 (4) |
| H19 | -0.0888 | 0.1447 | 0.9307 | 0.067* |
| C20 | 0.34155 (16) | 0.27631 (9) | 0.80821 (15) | 0.0539 (4) |
| H20A | 0.2550 | 0.2794 | 0.8196 | 0.081* |
| H20B | 0.3569 | 0.2995 | 0.7314 | 0.081* |
| H20C | 0.3882 | 0.3018 | 0.8747 | 0.081* |
| C21 | 0.52053 (15) | 0.19025 (12) | 0.79462 (18) | 0.0646 (5) |
| H21A | 0.5634 | 0.2149 | 0.8642 | 0.097* |
| H21B | 0.5386 | 0.2153 | 0.7202 | 0.097* |
| H21C | 0.5470 | 0.1389 | 0.7920 | 0.097* |
| C22 | 0.34321 (13) | 0.07146 (8) | 0.67344 (13) | 0.0397 (3) |
| C23 | 0.40686 (15) | 0.02446 (10) | 0.75922 (15) | 0.0532 (4) |
| H23 | 0.4300 | 0.0424 | 0.8381 | 0.064* |
| C24 | 0.43630 (17) | -0.04821 (10) | 0.72953 (19) | 0.0638 (5) |
| H24 | 0.4789 | -0.0787 | 0.7882 | 0.077* |
| C25 | 0.40313 (18) | -0.07557 (10) | 0.6143 (2) | 0.0700 (5) |
| H25 | 0.4234 | -0.1246 | 0.5943 | 0.084* |
| C26 | 0.33977 (17) | -0.03070 (10) | 0.52792 (18) | 0.0627 (5) |
| H26 | 0.3162 | -0.0495 | 0.4497 | 0.075* |
| C27 | 0.31101 (14) | 0.04226 (9) | 0.55720 (14) | 0.0475 (4) |
| H27 | 0.2691 | 0.0724 | 0.4976 | 0.057* |

Atomic displacement parameters (\AA^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|----|------------|-------------|------------|-------------|-------------|-------------|
| O1 | 0.0589 (7) | 0.0747 (8) | 0.0352 (5) | 0.0155 (6) | 0.0096 (5) | 0.0121 (5) |
| O2 | 0.0575 (8) | 0.1115 (11) | 0.0444 (6) | -0.0127 (7) | -0.0138 (5) | -0.0013 (6) |
| N1 | 0.0348 (6) | 0.0364 (6) | 0.0304 (5) | -0.0024 (4) | 0.0035 (4) | 0.0021 (4) |
| C2 | 0.0398 (7) | 0.0379 (7) | 0.0308 (6) | -0.0046 (6) | 0.0045 (5) | 0.0026 (5) |
| C3 | 0.0469 (9) | 0.0608 (9) | 0.0314 (6) | -0.0062 (7) | -0.0002 (6) | 0.0079 (6) |
| C4 | 0.0465 (9) | 0.0623 (10) | 0.0377 (7) | -0.0050 (7) | -0.0046 (6) | -0.0025 (6) |

| | | | | | | |
|-----|-------------|-------------|-------------|--------------|--------------|--------------|
| C5 | 0.0382 (8) | 0.0535 (8) | 0.0418 (7) | -0.0107 (6) | 0.0023 (6) | -0.0053 (6) |
| C6 | 0.0346 (7) | 0.0400 (7) | 0.0342 (6) | -0.0031 (5) | 0.0044 (5) | 0.0009 (5) |
| C7 | 0.0424 (8) | 0.0365 (7) | 0.0329 (6) | -0.0008 (6) | 0.0025 (5) | 0.0033 (5) |
| C8 | 0.0400 (8) | 0.0384 (7) | 0.0341 (6) | 0.0001 (6) | 0.0012 (5) | 0.0054 (5) |
| C9 | 0.0520 (10) | 0.0482 (9) | 0.0602 (9) | 0.0012 (7) | -0.0090 (7) | -0.0065 (7) |
| C10 | 0.0455 (10) | 0.0635 (11) | 0.0914 (13) | -0.0029 (8) | -0.0171 (9) | -0.0044 (10) |
| C11 | 0.0421 (10) | 0.0613 (11) | 0.0772 (11) | 0.0092 (8) | 0.0096 (8) | 0.0086 (9) |
| C12 | 0.0585 (11) | 0.0559 (10) | 0.0623 (10) | 0.0107 (8) | 0.0065 (8) | -0.0072 (8) |
| C13 | 0.0465 (9) | 0.0514 (9) | 0.0530 (8) | 0.0019 (7) | -0.0029 (7) | -0.0096 (7) |
| C14 | 0.0380 (8) | 0.0398 (7) | 0.0372 (6) | -0.0028 (6) | -0.0002 (5) | 0.0102 (5) |
| C15 | 0.0461 (9) | 0.0402 (8) | 0.0588 (9) | -0.0051 (6) | 0.0020 (7) | 0.0019 (6) |
| C16 | 0.0605 (11) | 0.0441 (9) | 0.0748 (11) | -0.0121 (8) | -0.0106 (9) | 0.0065 (8) |
| C17 | 0.0543 (11) | 0.0667 (12) | 0.0860 (13) | -0.0249 (9) | -0.0094 (9) | 0.0264 (10) |
| C18 | 0.0478 (11) | 0.0892 (15) | 0.0825 (13) | -0.0161 (10) | 0.0185 (9) | 0.0165 (11) |
| C19 | 0.0482 (10) | 0.0659 (11) | 0.0556 (9) | -0.0082 (8) | 0.0125 (7) | 0.0015 (8) |
| C20 | 0.0578 (10) | 0.0517 (9) | 0.0534 (9) | -0.0181 (8) | 0.0115 (7) | -0.0115 (7) |
| C21 | 0.0408 (10) | 0.0849 (13) | 0.0679 (11) | -0.0175 (9) | 0.0028 (8) | -0.0122 (9) |
| C22 | 0.0322 (7) | 0.0419 (7) | 0.0450 (7) | -0.0006 (6) | 0.0034 (5) | 0.0021 (6) |
| C23 | 0.0476 (9) | 0.0557 (9) | 0.0546 (9) | 0.0042 (7) | -0.0047 (7) | 0.0055 (7) |
| C24 | 0.0506 (10) | 0.0528 (10) | 0.0855 (13) | 0.0092 (8) | -0.0085 (9) | 0.0137 (9) |
| C25 | 0.0571 (12) | 0.0432 (9) | 0.1068 (15) | 0.0095 (8) | -0.0094 (10) | -0.0086 (10) |
| C26 | 0.0609 (11) | 0.0520 (10) | 0.0728 (11) | 0.0076 (8) | -0.0074 (9) | -0.0174 (8) |
| C27 | 0.0454 (9) | 0.0468 (8) | 0.0493 (8) | 0.0055 (7) | -0.0021 (6) | -0.0041 (6) |

Geometric parameters (Å, °)

| | | | |
|---------|-------------|----------|-----------|
| O1—C7 | 1.2307 (16) | C14—C19 | 1.382 (2) |
| O2—C4 | 1.2115 (17) | C14—C15 | 1.387 (2) |
| N1—C7 | 1.3561 (16) | C15—C16 | 1.383 (2) |
| N1—C6 | 1.4754 (17) | C15—H15 | 0.9300 |
| N1—C2 | 1.4813 (15) | C16—C17 | 1.378 (3) |
| C2—C14 | 1.5170 (19) | C16—H16 | 0.9300 |
| C2—C3 | 1.5359 (19) | C17—C18 | 1.362 (3) |
| C2—H2 | 0.9800 | C17—H17 | 0.9300 |
| C3—C4 | 1.508 (2) | C18—C19 | 1.384 (3) |
| C3—H3A | 0.9700 | C18—H18 | 0.9300 |
| C3—H3B | 0.9700 | C19—H19 | 0.9300 |
| C4—C5 | 1.513 (2) | C20—H20A | 0.9600 |
| C5—C21 | 1.531 (2) | C20—H20B | 0.9600 |
| C5—C6 | 1.5412 (19) | C20—H20C | 0.9600 |
| C5—C20 | 1.545 (2) | C21—H21A | 0.9600 |
| C6—C22 | 1.529 (2) | C21—H21B | 0.9600 |
| C6—H6 | 0.9800 | C21—H21C | 0.9600 |
| C7—C8 | 1.495 (2) | C22—C27 | 1.384 (2) |
| C8—C9 | 1.381 (2) | C22—C23 | 1.392 (2) |
| C8—C13 | 1.384 (2) | C23—C24 | 1.378 (2) |
| C9—C10 | 1.380 (3) | C23—H23 | 0.9300 |
| C9—H9 | 0.9300 | C24—C25 | 1.364 (3) |
| C10—C11 | 1.364 (3) | C24—H24 | 0.9300 |

supplementary materials

| | | | |
|-------------|-------------|---------------|-------------|
| C10—H10 | 0.9300 | C25—C26 | 1.372 (3) |
| C11—C12 | 1.367 (3) | C25—H25 | 0.9300 |
| C11—H11 | 0.9300 | C26—C27 | 1.380 (2) |
| C12—C13 | 1.378 (2) | C26—H26 | 0.9300 |
| C12—H12 | 0.9300 | C27—H27 | 0.9300 |
| C13—H13 | 0.9300 | | |
| C7—N1—C6 | 118.40 (10) | C8—C13—H13 | 119.7 |
| C7—N1—C2 | 120.99 (11) | C19—C14—C15 | 118.55 (14) |
| C6—N1—C2 | 120.31 (10) | C19—C14—C2 | 119.56 (13) |
| N1—C2—C14 | 112.00 (10) | C15—C14—C2 | 121.88 (13) |
| N1—C2—C3 | 110.59 (11) | C16—C15—C14 | 120.51 (16) |
| C14—C2—C3 | 110.06 (11) | C16—C15—H15 | 119.7 |
| N1—C2—H2 | 108.0 | C14—C15—H15 | 119.7 |
| C14—C2—H2 | 108.0 | C17—C16—C15 | 120.16 (18) |
| C3—C2—H2 | 108.0 | C17—C16—H16 | 119.9 |
| C4—C3—C2 | 116.94 (12) | C15—C16—H16 | 119.9 |
| C4—C3—H3A | 108.1 | C18—C17—C16 | 119.66 (17) |
| C2—C3—H3A | 108.1 | C18—C17—H17 | 120.2 |
| C4—C3—H3B | 108.1 | C16—C17—H17 | 120.2 |
| C2—C3—H3B | 108.1 | C17—C18—C19 | 120.68 (18) |
| H3A—C3—H3B | 107.3 | C17—C18—H18 | 119.7 |
| O2—C4—C3 | 120.83 (14) | C19—C18—H18 | 119.7 |
| O2—C4—C5 | 122.41 (15) | C14—C19—C18 | 120.43 (17) |
| C3—C4—C5 | 116.74 (12) | C14—C19—H19 | 119.8 |
| C4—C5—C21 | 113.12 (14) | C18—C19—H19 | 119.8 |
| C4—C5—C6 | 109.53 (12) | C5—C20—H20A | 109.5 |
| C21—C5—C6 | 111.08 (12) | C5—C20—H20B | 109.5 |
| C4—C5—C20 | 105.76 (12) | H20A—C20—H20B | 109.5 |
| C21—C5—C20 | 108.03 (13) | C5—C20—H20C | 109.5 |
| C6—C5—C20 | 109.10 (12) | H20A—C20—H20C | 109.5 |
| N1—C6—C22 | 112.86 (11) | H20B—C20—H20C | 109.5 |
| N1—C6—C5 | 109.16 (10) | C5—C21—H21A | 109.5 |
| C22—C6—C5 | 116.95 (12) | C5—C21—H21B | 109.5 |
| N1—C6—H6 | 105.6 | H21A—C21—H21B | 109.5 |
| C22—C6—H6 | 105.6 | C5—C21—H21C | 109.5 |
| C5—C6—H6 | 105.6 | H21A—C21—H21C | 109.5 |
| O1—C7—N1 | 121.65 (13) | H21B—C21—H21C | 109.5 |
| O1—C7—C8 | 118.71 (12) | C27—C22—C23 | 117.29 (14) |
| N1—C7—C8 | 119.61 (11) | C27—C22—C6 | 117.74 (12) |
| C9—C8—C13 | 118.61 (14) | C23—C22—C6 | 124.95 (13) |
| C9—C8—C7 | 119.78 (13) | C24—C23—C22 | 121.26 (16) |
| C13—C8—C7 | 121.19 (13) | C24—C23—H23 | 119.4 |
| C10—C9—C8 | 120.05 (16) | C22—C23—H23 | 119.4 |
| C10—C9—H9 | 120.0 | C25—C24—C23 | 120.19 (16) |
| C8—C9—H9 | 120.0 | C25—C24—H24 | 119.9 |
| C11—C10—C9 | 120.93 (17) | C23—C24—H24 | 119.9 |
| C11—C10—H10 | 119.5 | C24—C25—C26 | 119.90 (17) |
| C9—C10—H10 | 119.5 | C24—C25—H25 | 120.0 |
| C10—C11—C12 | 119.45 (16) | C26—C25—H25 | 120.0 |

| | | | |
|---------------|--------------|-----------------|--------------|
| C10—C11—H11 | 120.3 | C25—C26—C27 | 119.99 (17) |
| C12—C11—H11 | 120.3 | C25—C26—H26 | 120.0 |
| C11—C12—C13 | 120.44 (16) | C27—C26—H26 | 120.0 |
| C11—C12—H12 | 119.8 | C26—C27—C22 | 121.36 (15) |
| C13—C12—H12 | 119.8 | C26—C27—H27 | 119.3 |
| C12—C13—C8 | 120.50 (15) | C22—C27—H27 | 119.3 |
| C12—C13—H13 | 119.7 | | |
| C7—N1—C2—C14 | 61.05 (16) | C13—C8—C9—C10 | -0.2 (2) |
| C6—N1—C2—C14 | -125.24 (12) | C7—C8—C9—C10 | -172.85 (15) |
| C7—N1—C2—C3 | -175.79 (12) | C8—C9—C10—C11 | 1.3 (3) |
| C6—N1—C2—C3 | -2.08 (16) | C9—C10—C11—C12 | -1.1 (3) |
| N1—C2—C3—C4 | 40.28 (18) | C10—C11—C12—C13 | -0.2 (3) |
| C14—C2—C3—C4 | 164.56 (13) | C11—C12—C13—C8 | 1.3 (3) |
| C2—C3—C4—O2 | 154.70 (16) | C9—C8—C13—C12 | -1.1 (2) |
| C2—C3—C4—C5 | -26.3 (2) | C7—C8—C13—C12 | 171.44 (14) |
| O2—C4—C5—C21 | 30.2 (2) | N1—C2—C14—C19 | -144.35 (13) |
| C3—C4—C5—C21 | -148.72 (15) | C3—C2—C14—C19 | 92.20 (16) |
| O2—C4—C5—C6 | 154.74 (16) | N1—C2—C14—C15 | 36.78 (18) |
| C3—C4—C5—C6 | -24.22 (19) | C3—C2—C14—C15 | -86.67 (15) |
| O2—C4—C5—C20 | -87.8 (2) | C19—C14—C15—C16 | 0.1 (2) |
| C3—C4—C5—C20 | 93.23 (16) | C2—C14—C15—C16 | 178.96 (13) |
| C7—N1—C6—C22 | -102.18 (13) | C14—C15—C16—C17 | -0.3 (2) |
| C2—N1—C6—C22 | 83.95 (14) | C15—C16—C17—C18 | 0.0 (3) |
| C7—N1—C6—C5 | 125.98 (13) | C16—C17—C18—C19 | 0.4 (3) |
| C2—N1—C6—C5 | -47.89 (15) | C15—C14—C19—C18 | 0.3 (2) |
| C4—C5—C6—N1 | 59.99 (15) | C2—C14—C19—C18 | -178.58 (16) |
| C21—C5—C6—N1 | -174.33 (13) | C17—C18—C19—C14 | -0.6 (3) |
| C20—C5—C6—N1 | -55.34 (14) | N1—C6—C22—C27 | 72.51 (15) |
| C4—C5—C6—C22 | -69.65 (16) | C5—C6—C22—C27 | -159.62 (13) |
| C21—C5—C6—C22 | 56.03 (17) | N1—C6—C22—C23 | -108.93 (15) |
| C20—C5—C6—C22 | 175.02 (11) | C5—C6—C22—C23 | 18.9 (2) |
| C6—N1—C7—O1 | 16.05 (19) | C27—C22—C23—C24 | -0.1 (2) |
| C2—N1—C7—O1 | -170.12 (13) | C6—C22—C23—C24 | -178.70 (15) |
| C6—N1—C7—C8 | -162.10 (11) | C22—C23—C24—C25 | 0.0 (3) |
| C2—N1—C7—C8 | 11.73 (18) | C23—C24—C25—C26 | -0.3 (3) |
| O1—C7—C8—C9 | 59.09 (19) | C24—C25—C26—C27 | 0.8 (3) |
| N1—C7—C8—C9 | -122.69 (15) | C25—C26—C27—C22 | -1.0 (3) |
| O1—C7—C8—C13 | -113.39 (16) | C23—C22—C27—C26 | 0.6 (2) |
| N1—C7—C8—C13 | 64.83 (18) | C6—C22—C27—C26 | 179.29 (15) |

Hydrogen-bond geometry (Å, °)

| <i>D</i> —H... <i>A</i> | <i>D</i> —H | H... <i>A</i> | <i>D</i> ... <i>A</i> | <i>D</i> —H... <i>A</i> |
|-----------------------------|-------------|---------------|-----------------------|-------------------------|
| C6—H6...O1 | 0.98 | 2.29 | 2.7346 (17) | 106 |
| C2—H2...O1 ⁱ | 0.98 | 2.56 | 3.3784 (17) | 141 |
| C20—H20A...O1 ⁱ | 0.96 | 2.47 | 3.1885 (19) | 132 |
| C20—H20B...O2 ⁱⁱ | 0.96 | 2.52 | 3.470 (2) | 170 |

Symmetry codes: (i) *x*, -*y*+1/2, *z*+1/2; (ii) *x*, -*y*+1/2, *z*-1/2.

Fig. 1

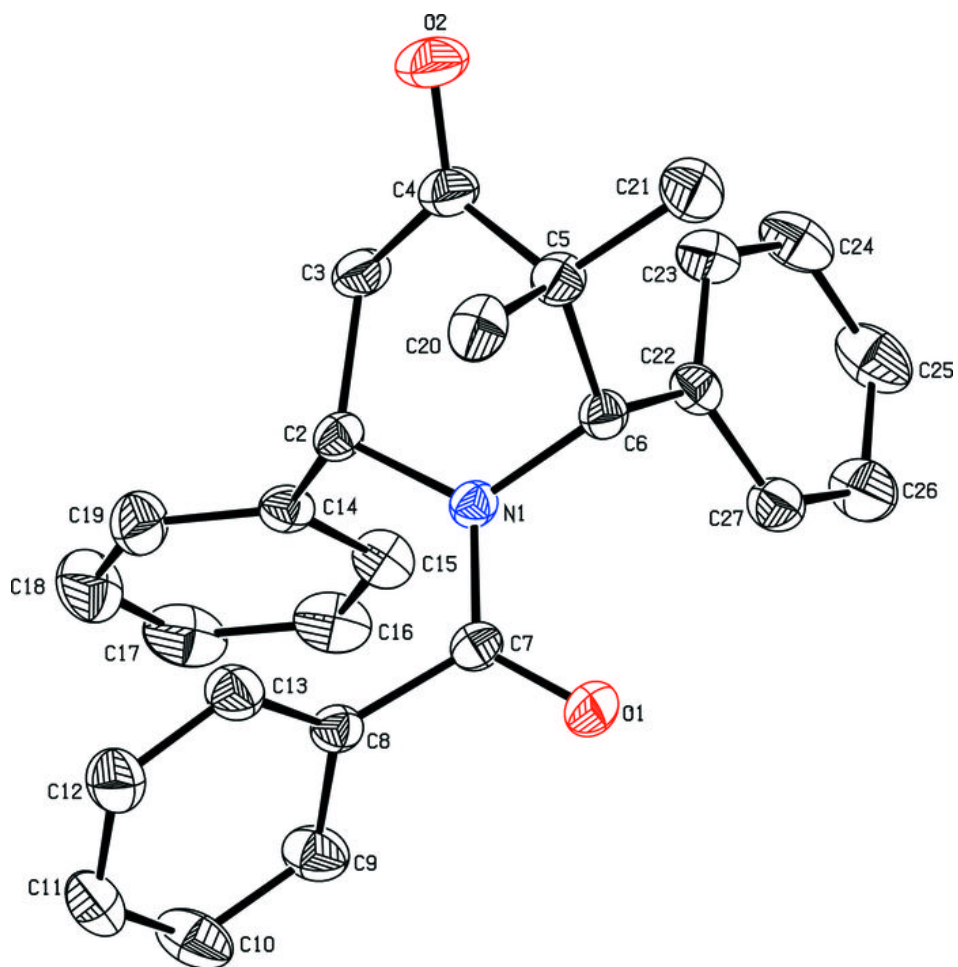


Fig. 2

